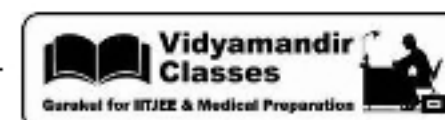




2

of 2



- 84.** The reaction of toluene with Cl_2 in presence of FeCl_3 gives predominantly : (2007)
- (A) benzoyl chloride (B) benzyl chloride
(C) o-and p-chlorotoluene (D) m-chlorotoluene
- 85.** Presence of a nitro group in a benzene ring : (2007)
- (A) activates the ring towards electrophilic substitution
(B) renders the ring basic
(C) deactivates the ring towards nucleophilic substitution
(D) deactivates the ring towards electrophilic substitution
- 86.** The compound formed as a result of oxidation of ethyl benzene by KMnO_4 is : (2007)
- (A) benzophenone (B) acetophenone
(C) benzoic acid (D) benzyl alcohol
- 87.** In the following sequence of reactions, the alkene affords the compound 'B'
- $$\text{CH}_3\text{CH}=\text{CHCH}_3 \xrightarrow{\text{O}_3} \text{A} \xrightarrow[\text{Zn}]{\text{H}_2\text{O}} \text{B}$$
- The compound B is : (2008)
- (A) $\text{CH}_3\text{CH}_2\text{CHO}$ (B) CH_3COCH_3
(C) $\text{CH}_3\text{CH}_2\text{COCH}_3$ (D) CH_3CHO
- 88.** The electrophile, E^+ attacks the benzene ring to generate the intermediate σ -complex. Of the following, which σ -complex is of lowest energy ? (2008)
- (A)

(B)

(C)

(D)
- 89.** The treatment of CH_3MgX with $\text{CH}_3\text{C}=\text{C}-\text{H}$ produces : (2008)
- (A) $\text{CH}_3-\text{CH}=\text{CH}_2$ (B) $\text{CH}_3\text{C}=\text{C}-\text{CH}_3$
(C) $\text{CH}_3-\overset{\text{H}}{\underset{|}{\text{C}}}=\overset{\text{H}}{\underset{|}{\text{C}}}-\text{CH}_3$ (D) CH_4
- 90.** The synthesis of 3-octyne is achieved by adding a bromoalkane into a mixture of sodium amide and alkyne. The bromoalkane and alkyne respectively are : (2010)
- (A) $\text{BrCH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$ and $\text{CH}_3\text{CH}_2\text{C}\equiv\text{CH}$
(B) $\text{BrCH}_2\text{CH}_2\text{CH}_3$ and $\text{CH}_3\text{CH}_2\text{CH}_2\text{C}\equiv\text{CH}$
(C) $\text{BrCH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$ and $\text{CH}_3\text{C}\equiv\text{CH}$
(D) $\text{BrCH}_2\text{CH}_2\text{CH}_2\text{CH}_3$ and $\text{CH}_3\text{CH}_2\text{C}\equiv\text{CH}$